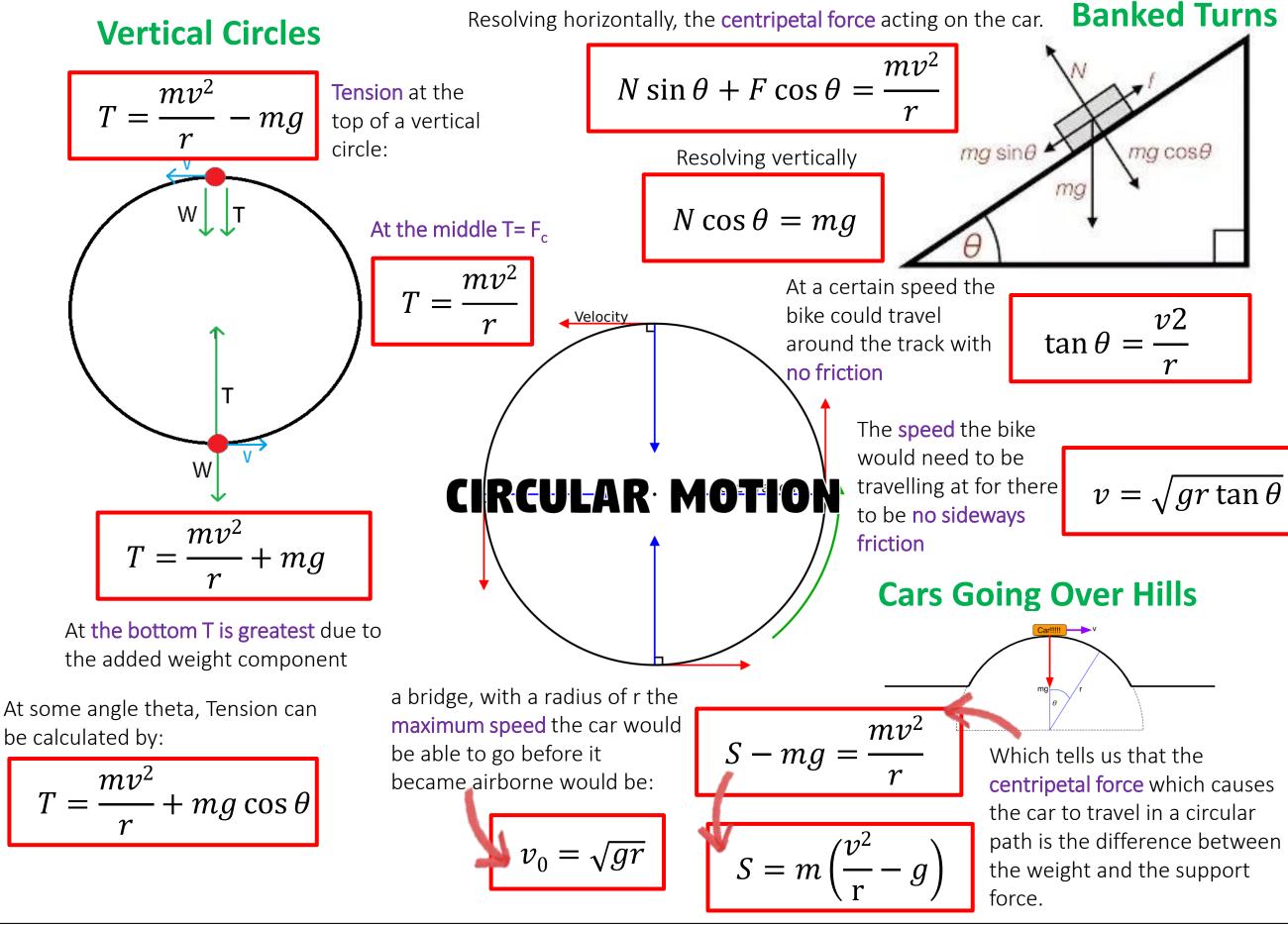


Blue equation-Given formulae

Red equation – Not given formulae



$$a = -\omega^2 x = -(2\pi f t)^2 x$$

For an object to undergo SHM, its acceleration must be directly proportional to its displacement from the equilibrium position in the opposite direction.

Phase Difference
$$=\frac{2\pi\Delta t}{t}=\omega\Delta t$$

$$a_{max} = \omega^2 A$$

Velocity vs displacement graph

velocity

The **acceleration** of the object will be at a maximum when it is at its maximum amplitude

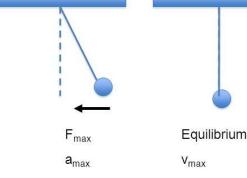
 $\mathsf{F}_{\mathsf{max}}$

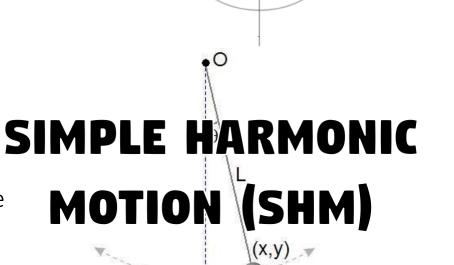
a_{max}

$$v = \omega \pm \sqrt{A^2 - x^2}$$

The velocity of the object will be at a maximum when it is passing through the equilibrium position

$$v_{max} = \omega A$$

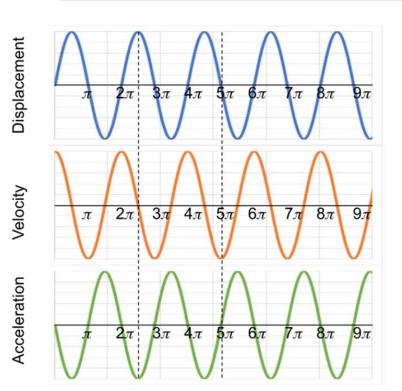




Fmax

a_{max}

displacemer



The gradient of the displacement graph will give you the velocity at any point.

 $x = A \cos \omega t$

The variation of

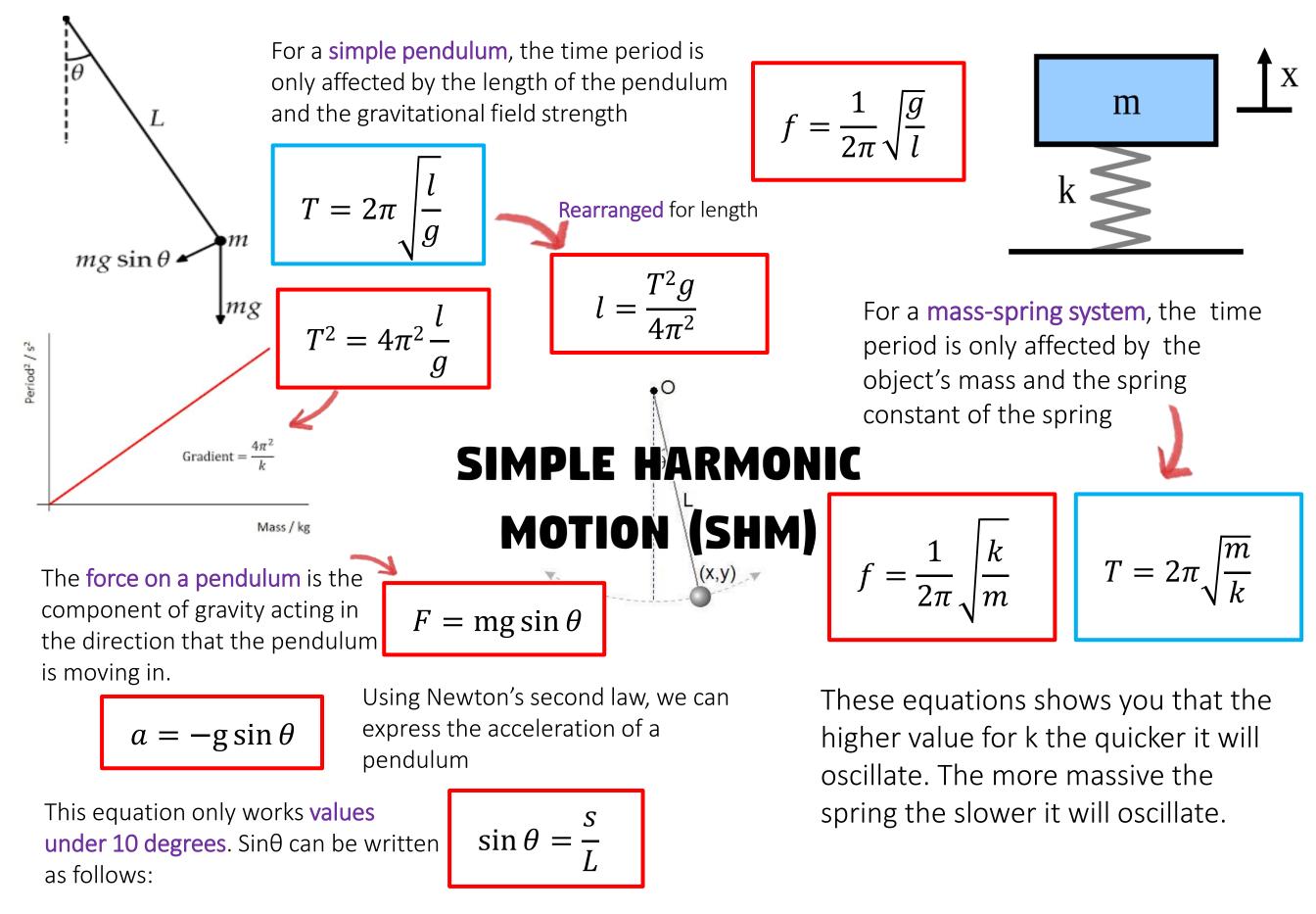
Equilibrium

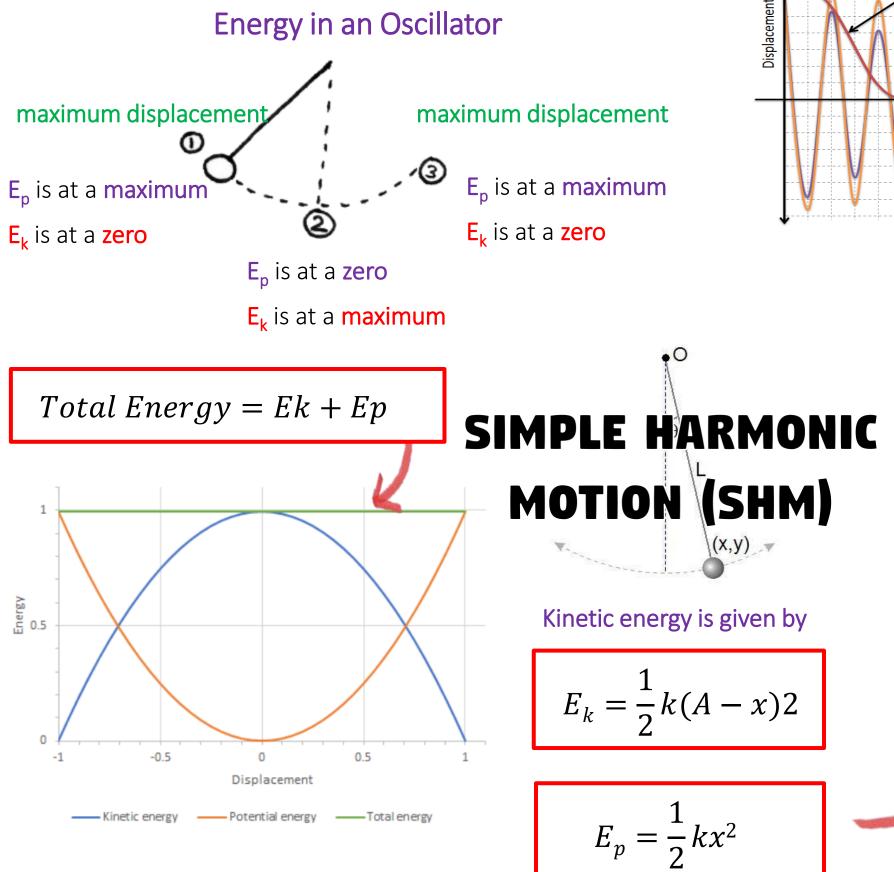
V_{max}

the displacement is is

$$x = A\cos 2\pi f t$$

At t = 0, the object is at maximum displacement. This is equal to the amplitude, A





Critical damping Light damping

> Light damping: oscillations of constant time period continue with the amplitudes gradually decreasing over time. Critical damping: the object returns to equilibrium in the shortest possible time without passing through to negative displacement.

Heavy damping: This results in the displaced object returning to equilibrium more slowly than if the system was critically damped.

$$E_{Total} = \frac{1}{2} k A^2$$

At the **maximum displacement**, the **total energy** of the system is

The potential energy in a spring is

$$K = C + 273$$

Absolute zero is the point at which all molecules have zero kinetic energy

A change of 1K equals a change of 1°C

 $\Delta U = \Delta Q - \Delta W$

Internal Energy

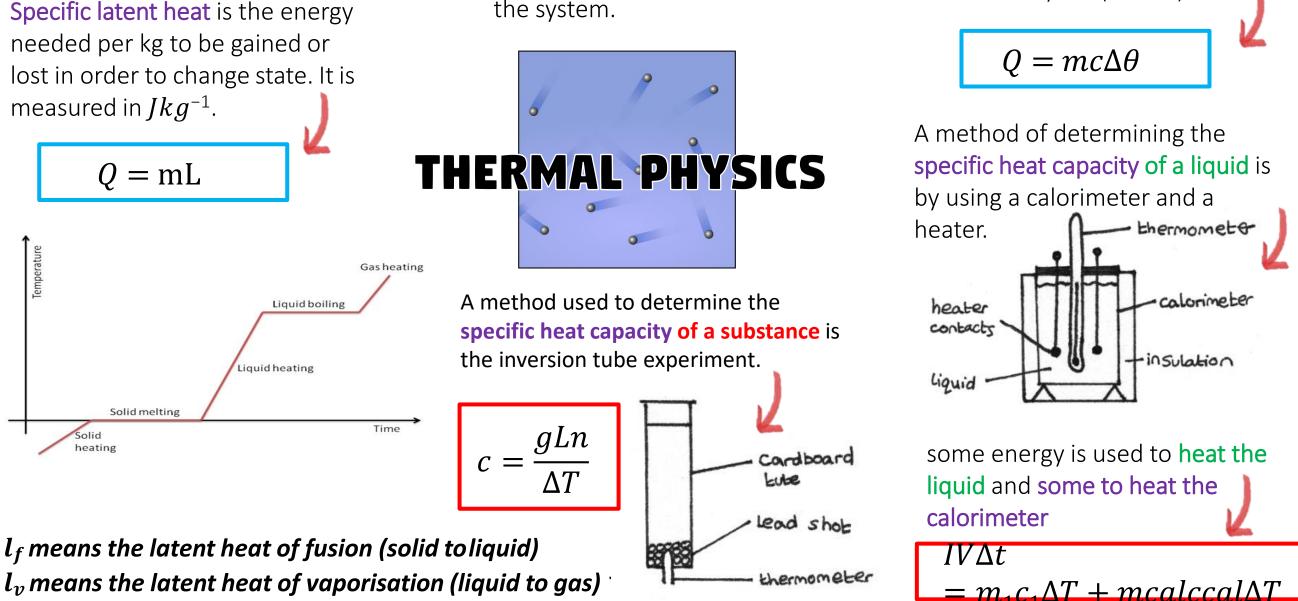
 $U = E_k - E_{pot}$

Specific heat capacity is the energy

of a substance by 1K (or $1^{\circ}C$).

needed to raise the temperature of 1kg

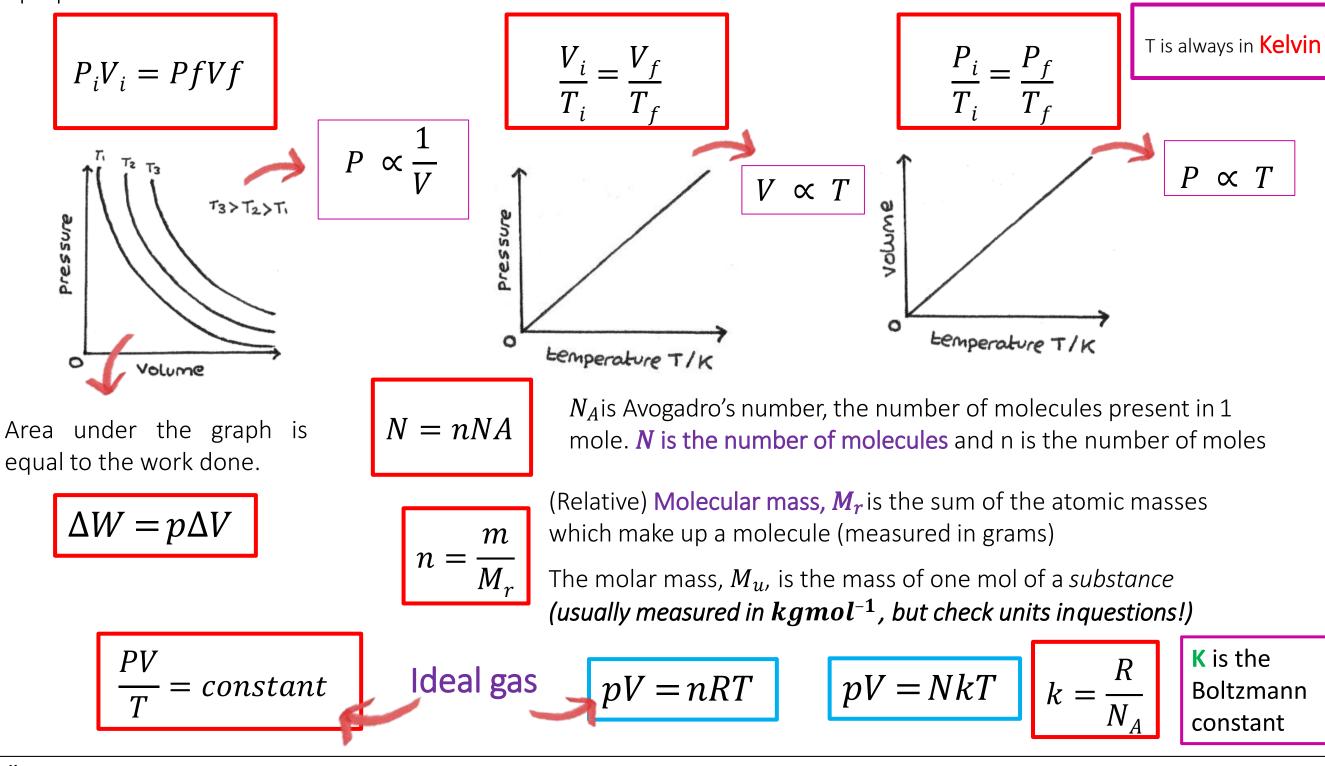
First law = The change in internal energy, ΔU , of a system is equal to the heat, ΔQ , added to the system minus the work done, ΔW , by the system.



Boyle's Law states that for a gas of fixed mass at a constant temperature, the pressure (p) and volume (V) are inversely proportional:

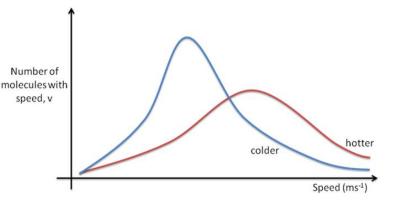
Charles's Law states that for a gas of fixed mass at constant pressure, the volume, V is directly proportional to its absolute temperature, T.

The pressure law states that for a gas of fixed mass at constant volume, the pressure, p is directly proportional to its absolute temperature, T.



Kinetic Theory and Molecular Speeds

Distribution of speed remains the same provided the temperature is constant.



Ideal Gas Assumptions

- \mathbf{R} The motion of all molecules is random.
- A No attraction between molecules
- V Molecules take up negligible volume.
- E All collisions between molecules and the walls of the container are completely elastic.

D – The duration of the collision is negligible in comparison to the duration between collisions

$$c_{rms} = \left[\frac{(C_1^2 + c_2^2 + c_3^2 \dots)}{N}\right]^{1/2}$$

T always in Kelvin

The rms speed of molecules in an ideal gas, this gives a mean of the magnitude of the speeds:

$$pV = \frac{1}{3} Nm(C_{rms})^2$$

The kinetic theory equation demonstrates that that for N molecules of gas at a given volume, **SICS** the mass and the r.m.s. speed affect the

pressure

$$E_{Kin} = \frac{3}{2}KT$$

macroscopic

Mean kinetic energy of one molecu

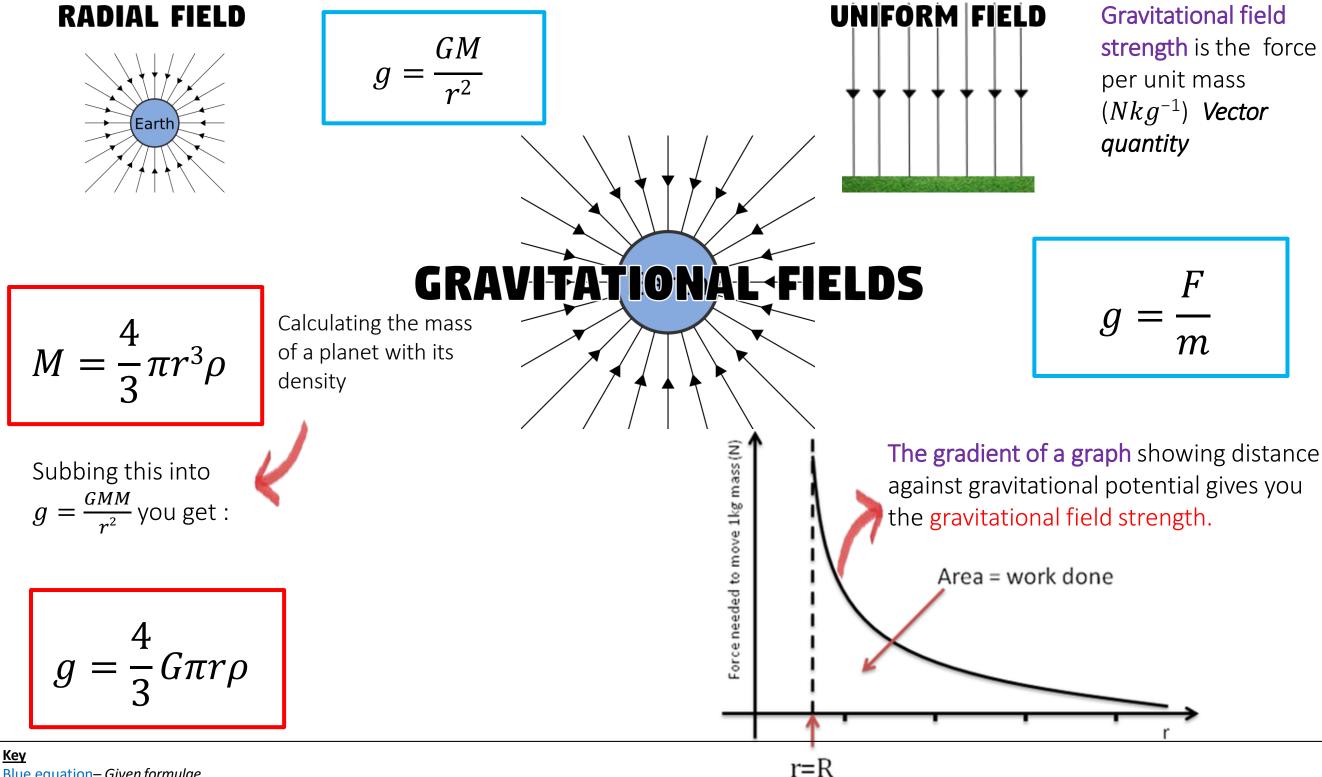
The total energy for n moles of an ideal gas

$$E_{K} = \frac{1}{2}m(c_{rms})^{2}$$
 $E_{KTotal} = \frac{3}{2}NKT$ $E_{K} = \frac{3}{2}nRT$

$$F_g = \frac{Gm_1m_2}{r^2}$$

Coulombs Law: Force between two masses is directly proportional to the product of their masses and inversely proportional to the distance between them squared. *Vector quantity*

 $\Delta E_{pot} = mg\Delta \mathbf{h}$



Gravitational Potential is:

the work done per unit mass in bringing a point mass from infinity to a point in a gravitational field (Jkg^{-1}) Scalar quantity

 $\Delta E_{\text{pot}} = \Delta W = m \Delta V$

Gravitational Potential is always negative as by convention, zero potential is at a point which is an infinite distance away from a gravitational field

Distance from the centre of the Earth

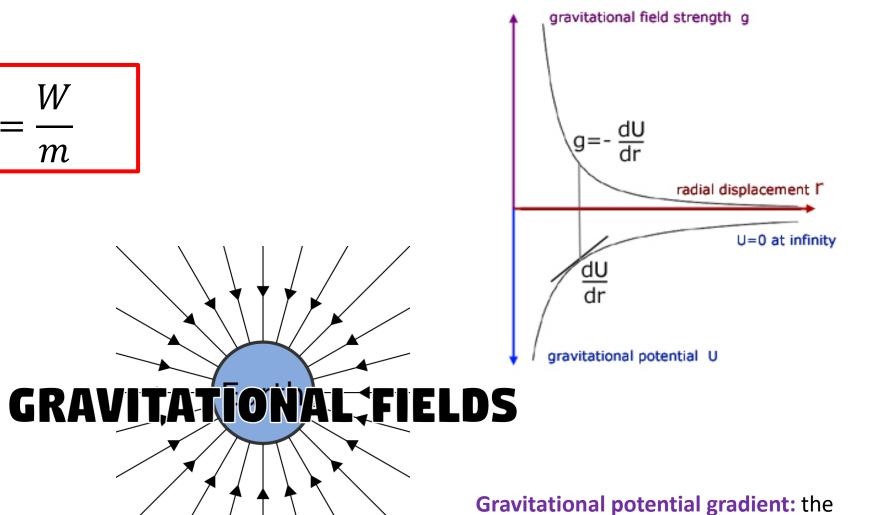
GM

R

$$\Delta V = \left(-GM\left(\frac{1}{r_1} - \frac{1}{r_2}\right)\right)$$

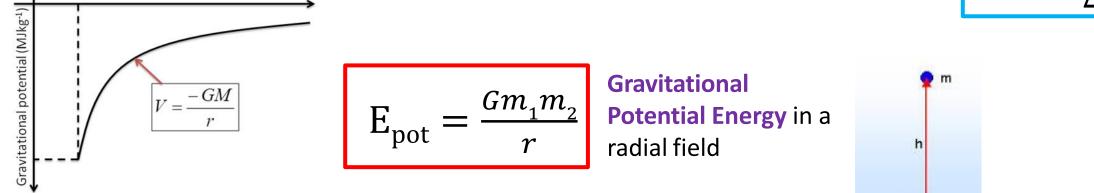
W

m



change of potential per metre at that point $\left(\frac{\Delta V}{\Lambda R}\right)$.

$$g = \frac{\Delta V}{\Delta R}$$



<u>Key</u> Blue equation-Given formulae Red equation – Not given formulae

 Gm_1m_2 2r

GM

Total energy INPUT required to put a satellite into an orbit of radius r around a planet of mass M and radius R is therefore the sum of the gravitational potential energy and the kinetic energy of the satellite

The orbital speed of a satellite

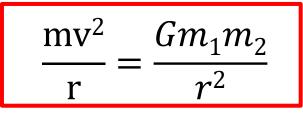
is inversely proportional to the

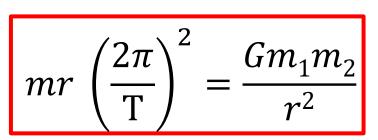
square root of the radius of its

R

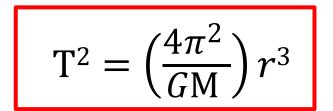
FIONAL

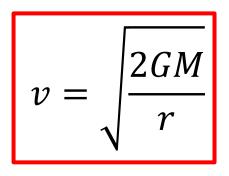
Centripetal Force = Gravitational Force



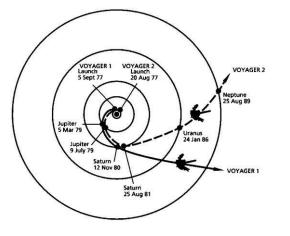


The **time period** of a satellite's orbit around a planet squared is directly to proportional to the radius of its orbit cubed (Circular motion applies)





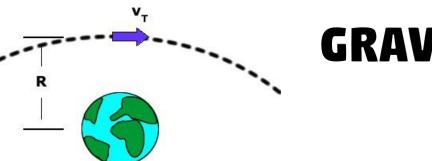
The escape velocity of an object is the velocity needed for an object to be completely free of a gravitational field from a planet



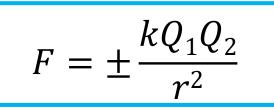
$$r^3 = \left(\frac{GM}{4\pi^2}\right)T^2$$

$$r^3 \propto T^2$$

<u>Key</u> Blue equation – *Given formulae* Red equation – *Not given formulae*

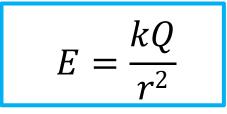


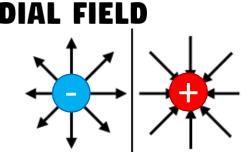
orbit

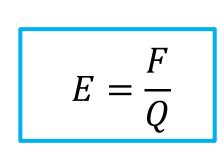


Coulombs Law: Force between two charges is directly proportional to the product of their charges and inversely proportional to the distance between them squared. Vector quantity,

RADIAL FIELD







ELECTRICAL FIELDS

 Q_1

 Q_{2}

х

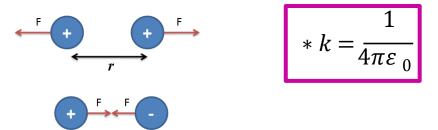
D

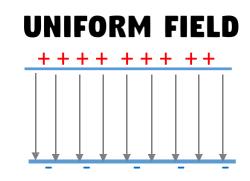
Electric field strength in a radial field is inversely proportional to the distance from the charge squared

 $\Delta W = Q \Delta V$

Electric Potential is the work done per unit charge in bringing a positive charge to infinity from a point in an electric field High potential $(IC^{-1} or V)$ Scalar quantity situation

Depending on the charge, electric potential may be positive or negative as it is the work done in bringing a positive point charge to a point in a field

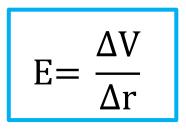




Electric field strength is the force per unit charge (NC^{-1}) Vector quantity

The electric field strength, E, is equal to the negative of the potential gradient

 $E_1 - E_2 = k$

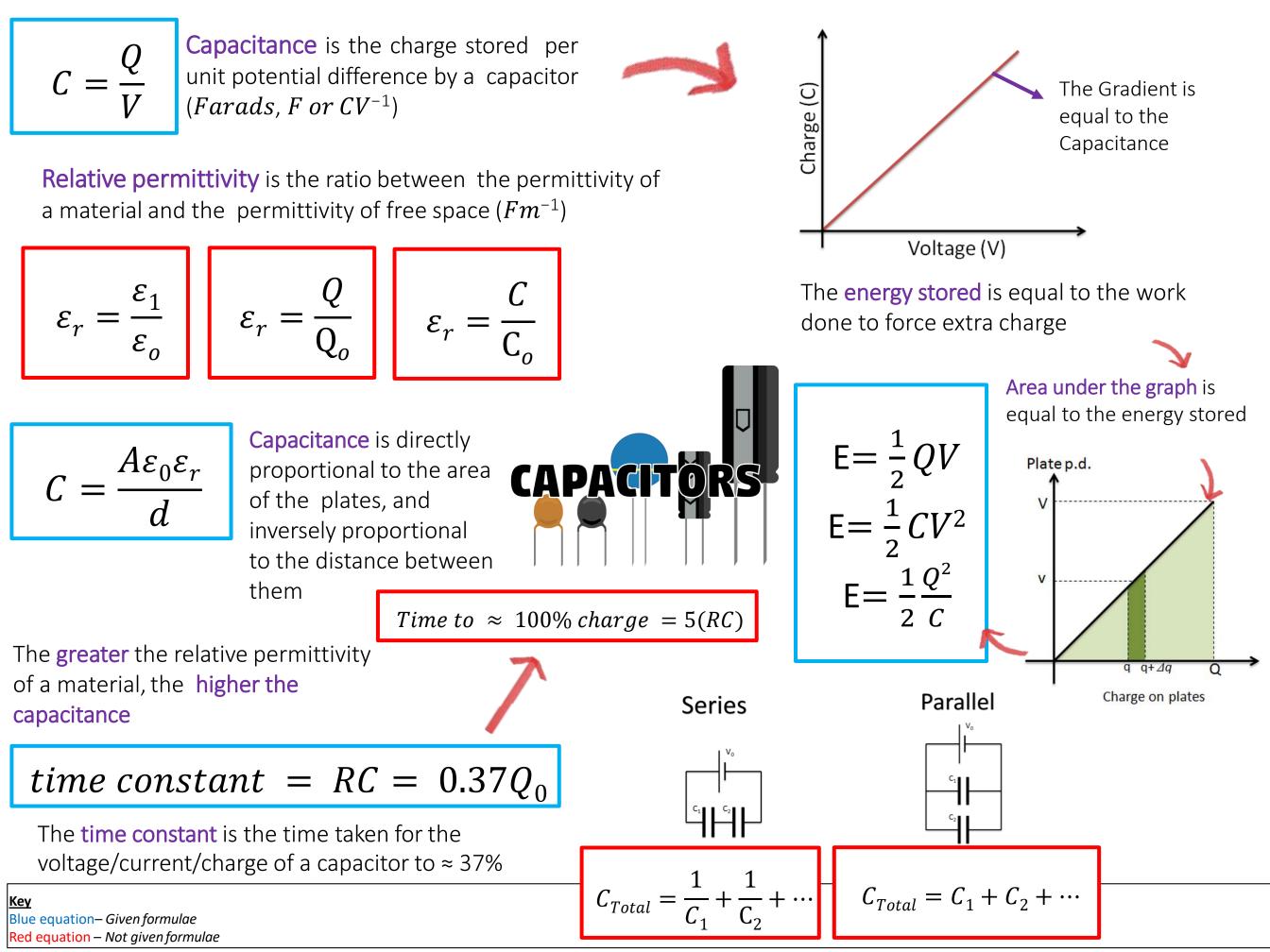


Electric fields when there is more than one charge

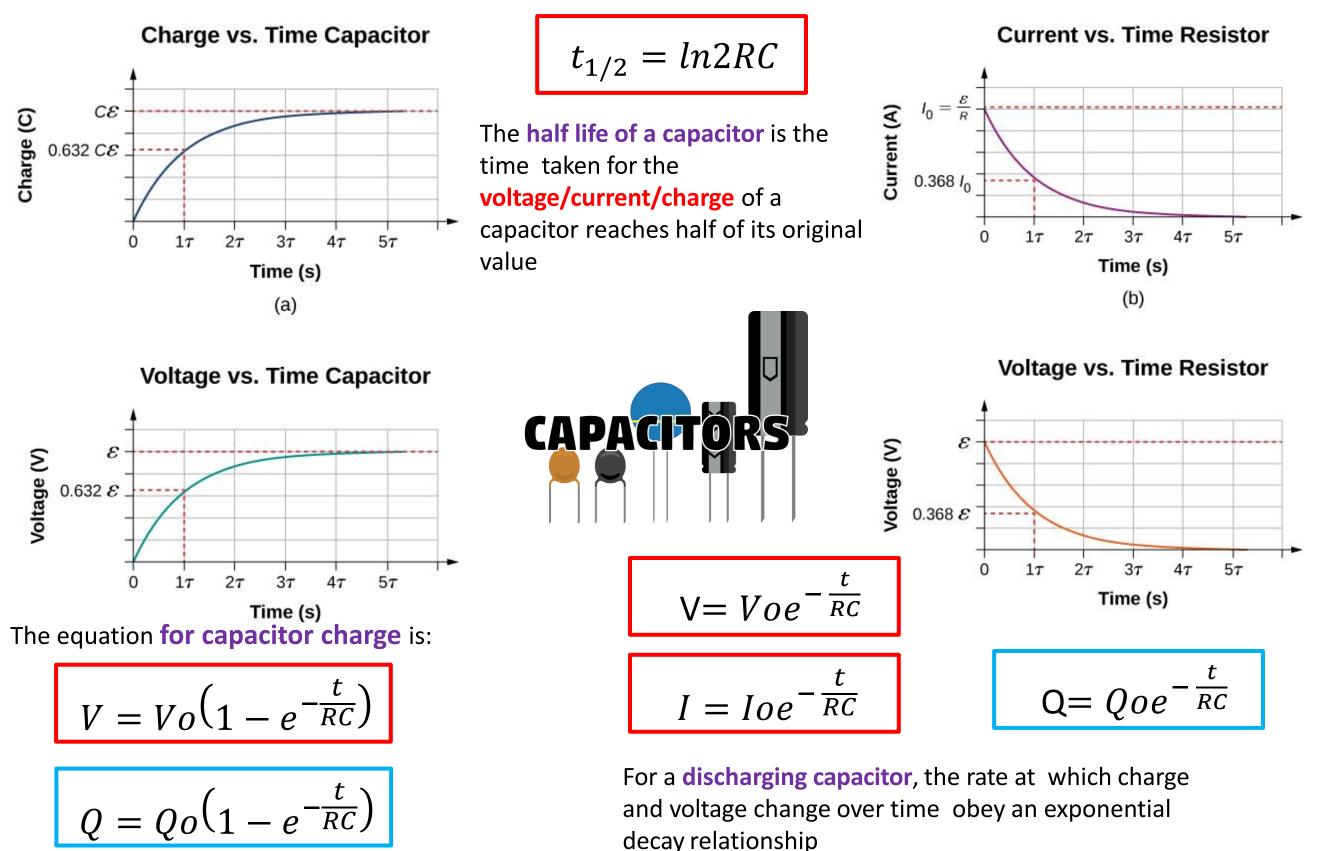
$$E_1 - E_2 = k \left(\frac{Q_1}{r_1^2} - \frac{Q_2}{r_2^2} \right)$$

Key Blue equation-Given formulae Red equation – Not given formulae

 $V = \pm$

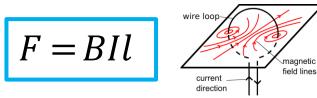


CHARGING CAPACITORS



Key Blue equation– Given formulae Red equation – Not given formulae

DISCHARGING CAPACITORS



F = (BIl)n

The **force on a current carrying** wire is directly proportional to the magnetic flux density (field strength), the current in the wire and the length of the wire.

F = Bev

The force on a charge in a magnetic field is affected by the magnetic flux density, the size of the charge and the velocity of its motion perpendicular to the field.

> Particle injected into the centre and exits here.

Alternating electric field

accelerates charge across

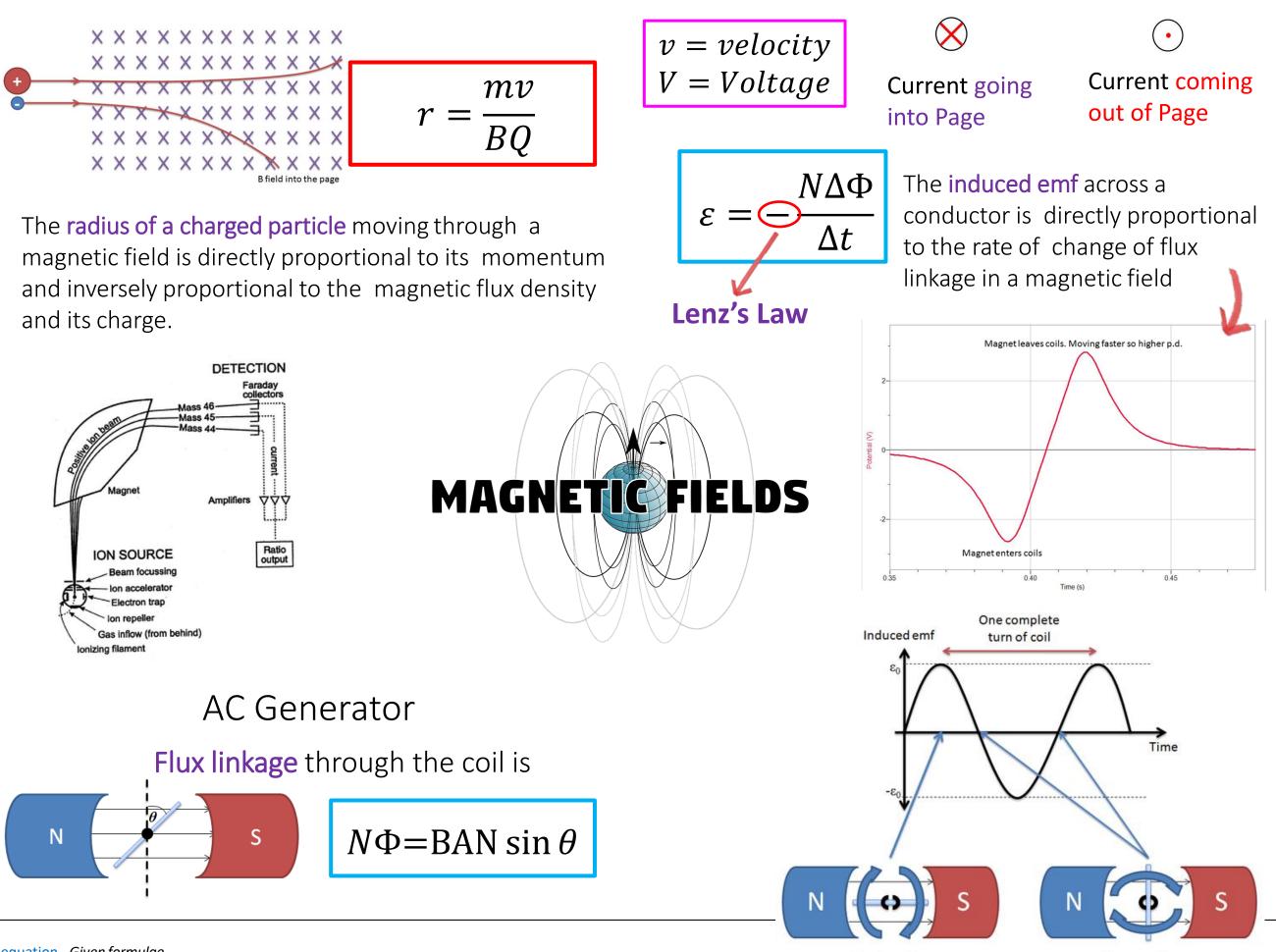
the gap.

 $2\pi m$

v = velocityMagnetic Flux Density is measured in $\Phi = BA$ V = Voltage*Teslas, T or* $NA^{-1}m^{-1}$ *Vector quantity* Magnetic Flux is the total magnetic flux passing through a given area. It is measured in Webers, *Wb or* Tm^{-2} *Scalar quantity* Path of charged Particle is circular from **MAGNETIC** FIELDS continuous B field into the diagram (green region). $N\Phi = BAN \cos \theta$ The emf of a coil rotating uniformly in a ε =BAN $\omega \sin \theta \omega t$ magnetic field is dependant on the flux linkage and the angular speed. Fleming's left hand rule velocity. Applied for charges moving through conventional current $+ve \rightarrow -ve$

The **time period** of a charged particle moving through a magnetic field is **independent of its velocity**.

°U~

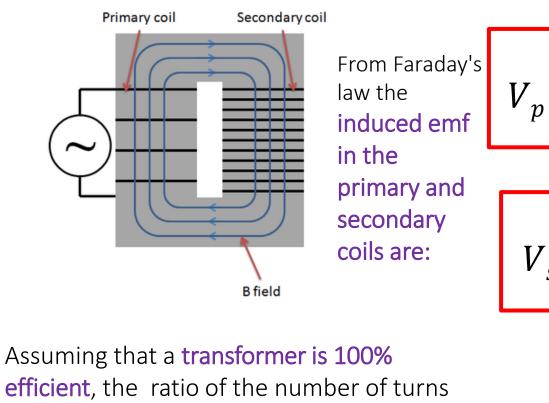


Blue equation – *Given formulae* Red equation – *Not given formulae*

Key

Zero voltage

Peak voltage



 $\frac{N_s}{N_p} = \frac{V_s}{V_p} \qquad \qquad \frac{N_s}{N_p} = \frac{V_s}{V_p}$

on the primary and secondary coil is

equal to the ratio of their voltages

$$Efficiency = \frac{I_s V_s}{I_p V_p} x100$$

$$\frac{\Delta \Phi}{\Delta t}$$

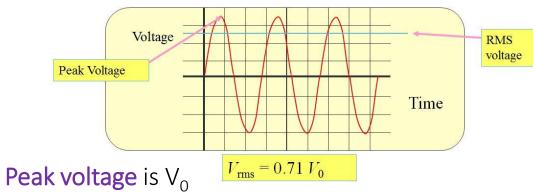
$$V_s = Ns \frac{\Delta \Phi}{\Delta t}$$

MAGNETIC FIELDS

Р

= *IrmsVrms*

 $I_p V_p = I_s V_s$

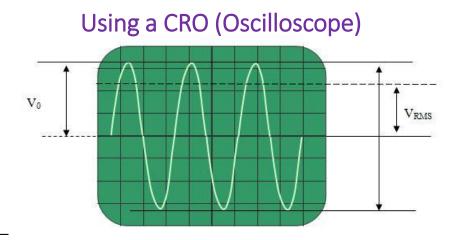


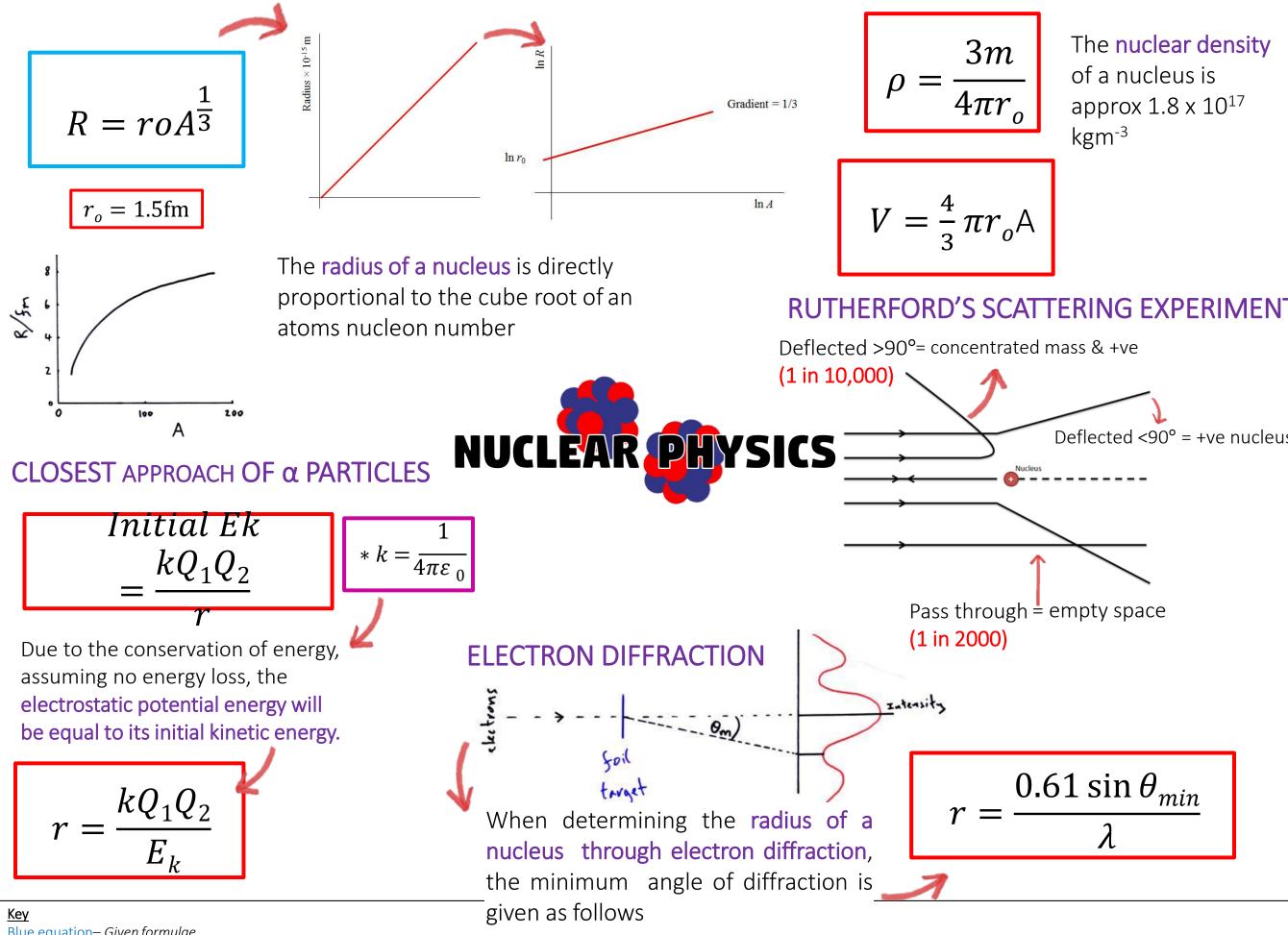
The **r.m.s. current/voltage** of an A.C supply can be directly compared to its DC equivalent

$$I_{rms} = \frac{I_0}{\sqrt{2}} \qquad V_{rr}$$

 $V_{rms} = \frac{V_0}{\sqrt{2}}$

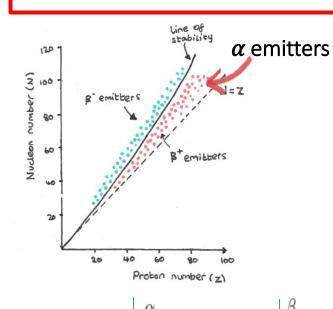
We use the rms value because its use allows us to do electrical calculations as if they were direct currents.





Alpha Decay, α Nucleon number decreases by 4 Proton number decreases by 2

$$X^A_Z \rightarrow X'^{A-4}_{Z-2} + \alpha^4_2$$



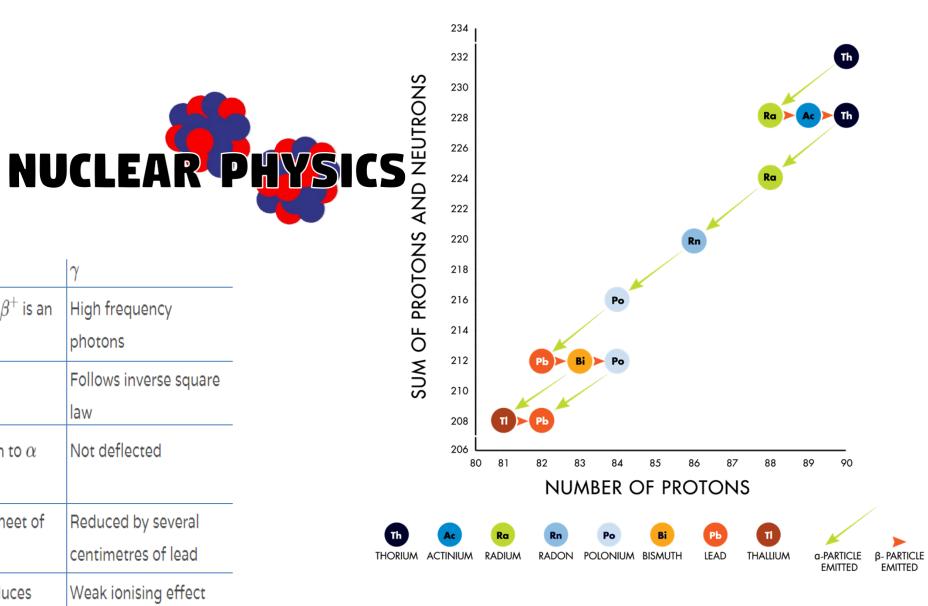
	α	β	γ
Description	2 protons and 2	eta^- is an electron. eta^+ is an	High frequency
	neutrons	positron.	photons
Range in air	<10cm	<1m	Follows inverse square
			law
Deflection in	Easily deflected	Opposite direction to $lpha$	Not deflected
magnetic field		and less easily	
Absorption	Stopped by paper	Stopped by thin sheet of	Reduced by several
		aluminium	centimetres of lead
lonisation	Intense, produces	Less intense, produces	Weak ionising effect
	10^4 ions per mm	about 100 ions per mm	

Beta minus decay, β^- Nucleon number stays the same Proton number increases by 1

$$X^A_Z \to X'^A_{Z+1} + \beta^{-0}_{-1} + \overline{V}_e$$

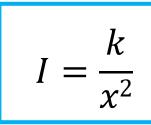
Beta plus decay, β^+ Nucleon number stays the same Proton number decreases by 1

$$X^A_Z \rightarrow X'^A_{Z-1} + \beta^{+\,0}_{-1} + ve$$



RADIOACTIVE EMISSION

Key



The intensity of radiation is inversely proportional to the distance from the source squared Nuclear decay is random and spontaneous. The decay constant, λ is the probability of a nucleus decaying in a given

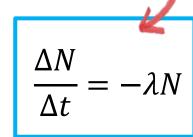
> time.

$$1u = 931.5MeV = 1.661 \times 10^{-27} kg$$

1 atomic mass unit has a binding energy of **931.5***MeV*

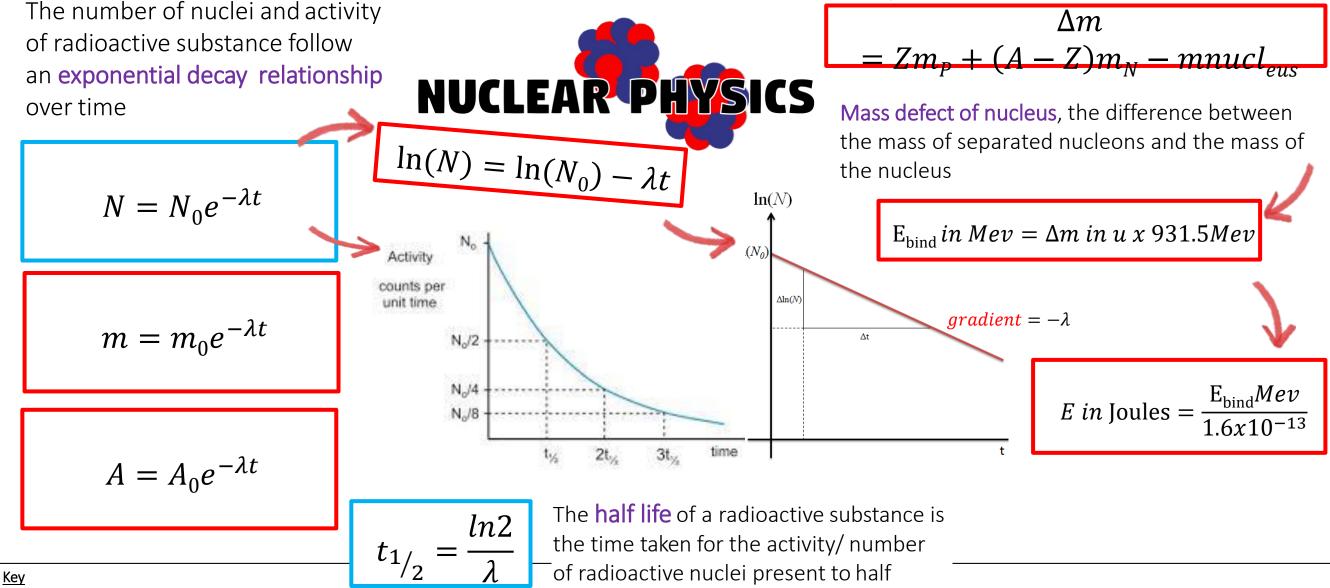
$$A = N\lambda$$

The **activity**, *A* of a radioactive sample is directly proportional to the number of nuclei present. It is measured in becquerels, *Bq*



$$E_{bind} = mc^2$$

The **binding energy** of a nucleus is the work that must be done to separate a nucleus into its constituent neutrons and protons.

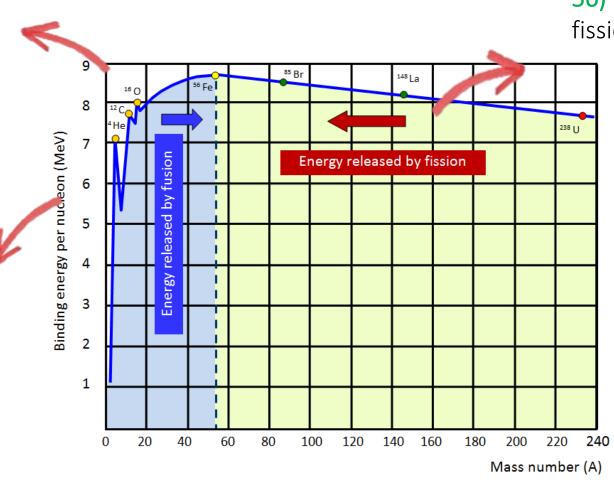




Nucleons with a mass number less than the peak (Fe-56) release energy through nuclear fusion

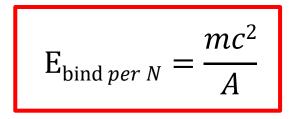
Fusion: two light nuclei fuse together to form a larger, more stable nucleus

> Oxygen-16, Carbon -12 and He-4 are very stable isotopes



Nucleons with a mass number greater than the peak release energy (Fe-56) through nuclear fission.

> Fission: In this process, a large unstable nucleus is split into two lighter, more stable nuclei.



The **binding energy per nucleon** is the **average work**

done per nucleon to remove all the nucleons from a nucleus. It is a measure of the stability of a nucleus.

Neutrons have ZERO binding energy

$$\frac{1}{f} = \frac{1}{u} + \frac{1}{V}$$

$$\frac{1}{r} = \frac{1}{u} + \frac{1}{V}$$

$$\frac{1}{r} = \frac{1}{u} + \frac{1}{V}$$

$$\frac{1}{r} = \frac{1}{r}$$

$$\frac{1}{r} = \frac{1}$$

MEDICAL PHYSICS

 $\mu_m =$

intensity level =
$$10\log \frac{I}{I_0}$$

The intensity of sound can also be measured in decibels (dB). The intensity level follows a logarithmic scale. This is for going from intensity in Wm^{-2} to decibels

$$I = I_0 \, x \, 10^{0.1(dB)}$$

Converting decibels back to Wm⁻²

$$\Delta L \propto \log\left(\frac{I_2}{I_1}\right)$$

Perceived loudness in decibels is proportional logarithmically to the ratio of intensity before and after

$$\frac{1}{T_E} = \frac{1}{T_B} + \frac{1}{T_P}$$

follows an exponential decay relationship

The half value thickness is the thickness of a material needed to reduce the intensity of an X-ray beam to half its original value

The mass attenuation coefficient is a measure of how much radiation is absorbed per unit mass of a material

Ln2

μ

 $x_{1/2}$

The effective half life of a radioactive

tracer depends on the physical half life of the substance and the biological half life which is how long the body's biological processes take to remove the substance